

Exploring the cause of source water contamination using LC/MS/MS



Nov. 26, 2015, UT



Mari ASAMI, Dr. Eng.

Chief Senior Researcher
Water Management Field
Department of Environmental Health
National Institute of Public Health

Professor

RECWET, the University of Tokyo



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Development of analytical method for LC(IC)/MS/MS

- We have developed analytical methods, conduct first national survey and did risk assessment respectively on **Bromate(IC, IC/MS/MS and LC/MS/MS), perchlorate(IC/MS/MS and LC/MS/MS), chlorate(IC), and NDMA(SPE-LC/MS/MS)**
- Collect water quality accidents related to health issues
- Contribution to guidelines for water safety plans



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Research topics

Risk management and chemical analysis in water source and drinking water using LC/MS/MS



Research promotion

Reflection to risk management



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Control of toxic disinfection by-products (DBPs)

Production of carcinogenic compound, bromate in ozonation as advanced water treatment

Development of analytical method of bromate
National survey

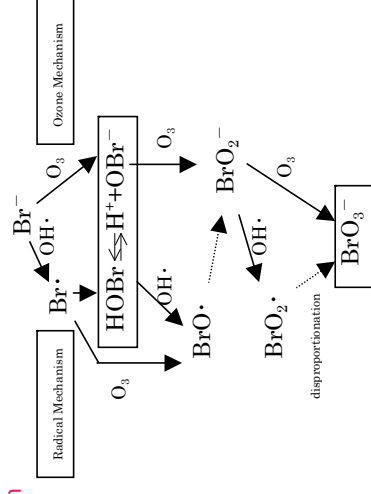
High concentration of bromate formation in cities of high bromide, treatment at high pH, high dose of ozone, and high temperature

Control method

Bromate was introduced into WQ standard, controlled by pH and ozone dose.



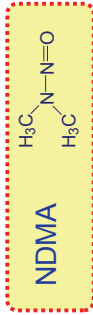
Major information introduced in advanced water treatment systems



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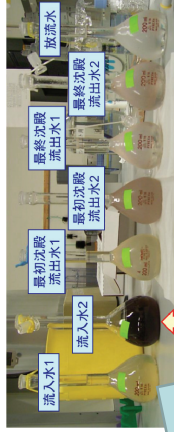
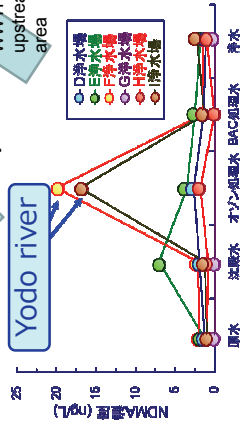
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Identification of toxic NDMA (1/2)

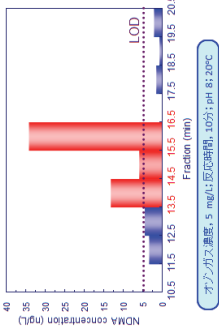


- 10⁻⁵ cancer potency (IRIS): 7 ng/L
- WHO GDWQ: 100 ng/L
- Japanese ref.: 100 ng/L

Development of analytical method for NDMA National survey



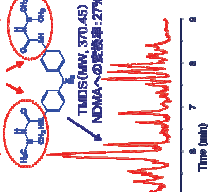
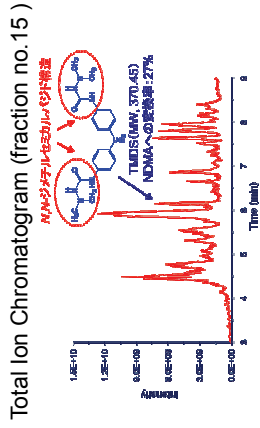
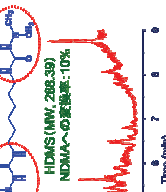
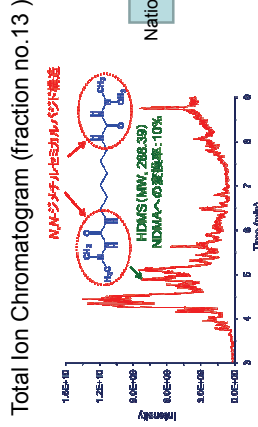
Looked for inflow of WWTPs in upstream area



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Identification of toxic NDMA (2/2)

National survey of HDMS and TMDS concentrations



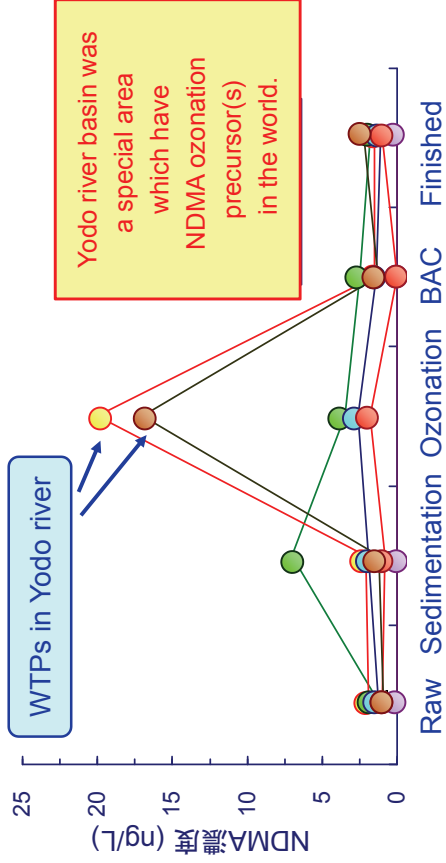
Identification of Antiproliferating Agents as Precursors of N-Nitrosodimethylamine Production on Disinfection from Sewage Treatment Plant Effluent

YOSHIZUMI, MITSUAKI, Y. AND HOSHINO, T. (2009) Environmental Health Perspectives, Special Issue of Environmental Health Perspectives, 117(10), 1533-1537

Candidates for "Troubling" substances for water treatment"

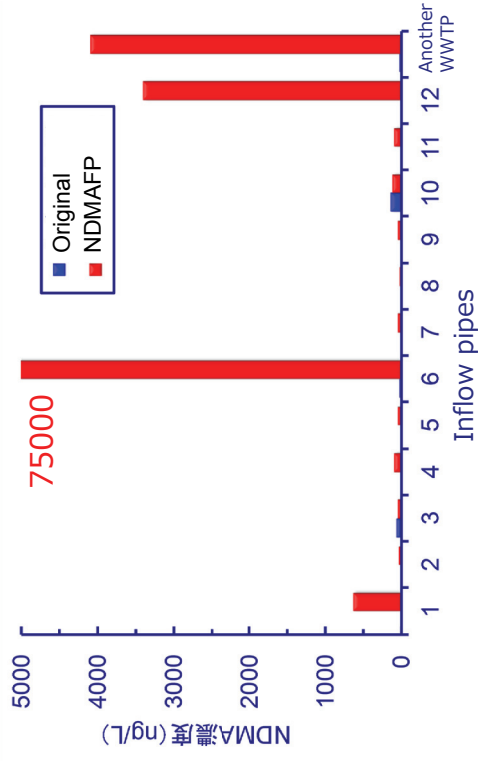
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NDMA behavior in ozonation and AC process



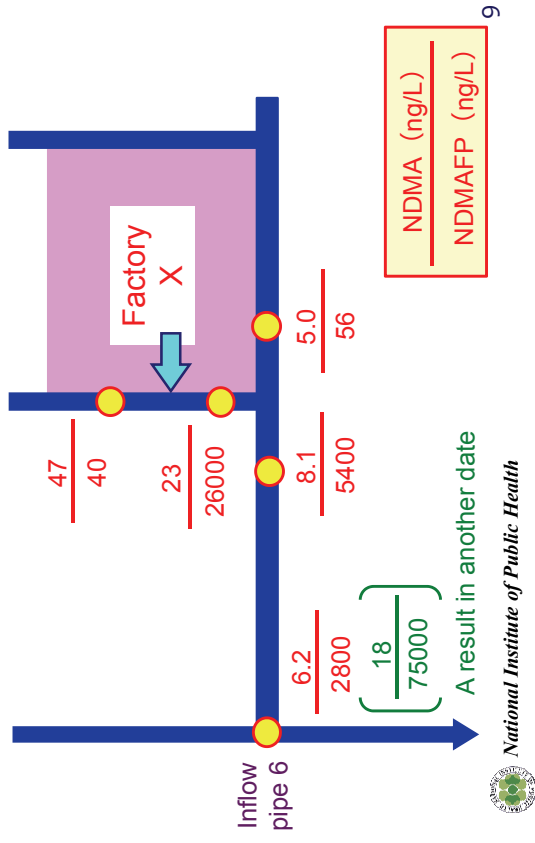
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NDMA and its formation potentials (NDMAFP) at inflow pipes of the A Waste water treatment plant



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NDMA and NDMAFP at inflow pipes in upstream



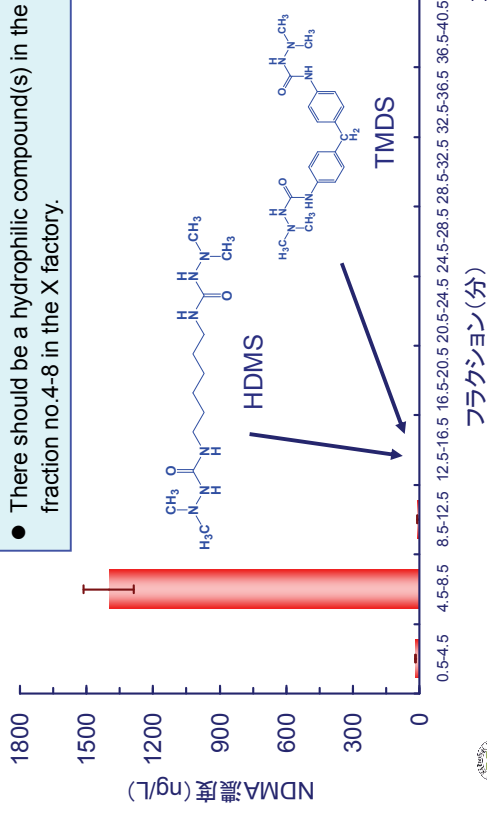
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Ozonation NDMAFP

Another component in the fraction no.4-8 in the discharge form the factory X.

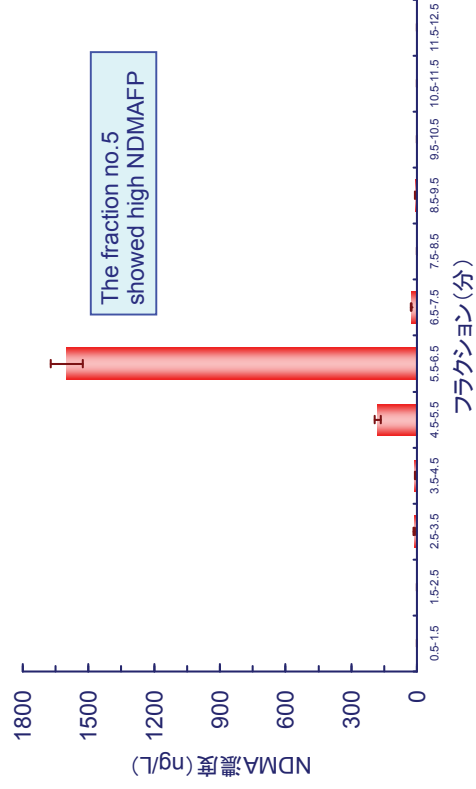
- There should be a hydrophilic compound(s) in the fraction no.4-8 in the X factory.



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The fraction no.5 showed high "Ozonation NDMAFP" in the discharge from the factory A.

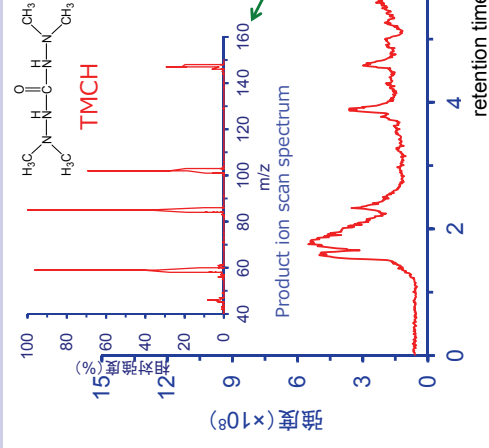


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UPLC-MS of Total Ion Chromatogram fraction no.5

- Retention time and spectrum of UPLC-MS/MS
- Accurate mass analysis using TOF-MS
- Peak in the fraction : 147.1246
- Synthesized TMCH : 147.1243
- ¹H-NMR / ¹³C-NMR



Identification of TMCH (1,1,5,5-tetramethyl carbonohydrazid)



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Exposure analysis of hydrophilic chemicals in water and foods

Data of chlorate and perchlorate in food was not available due to difficulties in analysis



Developed analytical method using isotope and LC/MS/MS



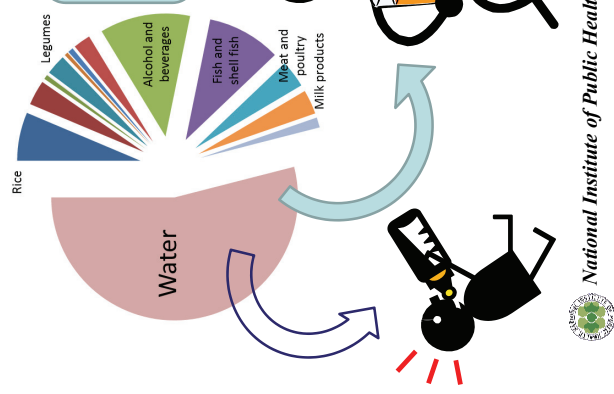
Exposure (intake) survey for each chlorate and perchlorate for each food groups using market –basket method



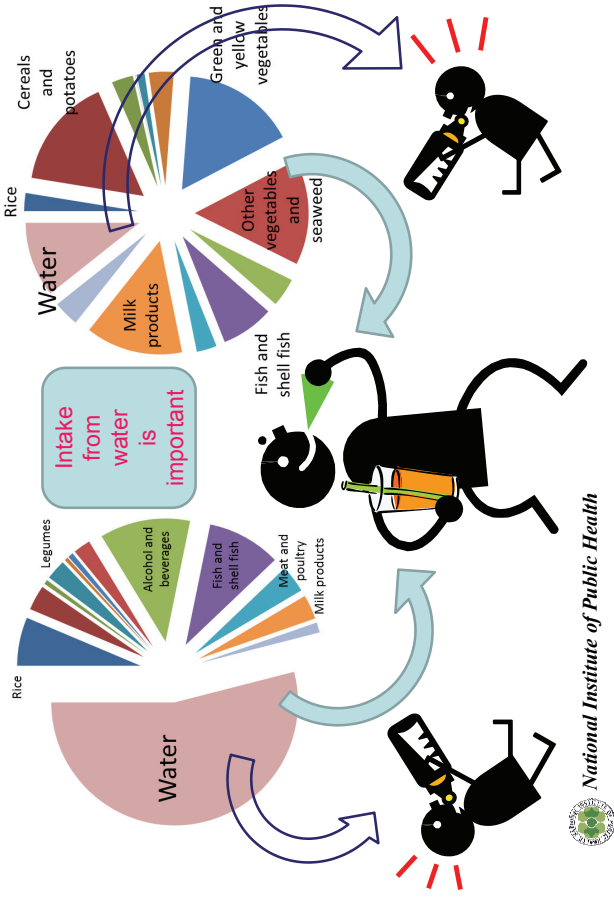
High contribution of water supply



Total intake of chlorate



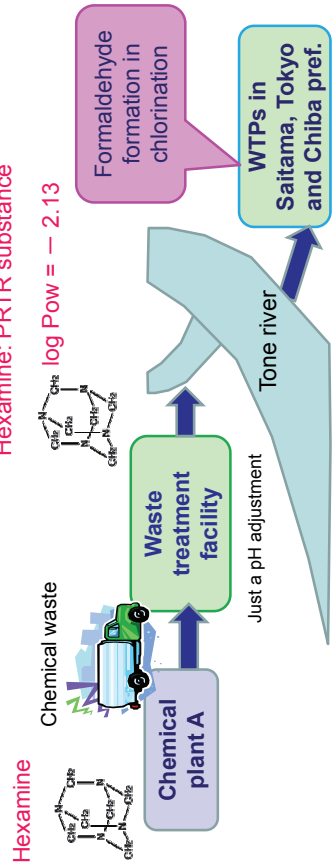
Total intake of perchlorate



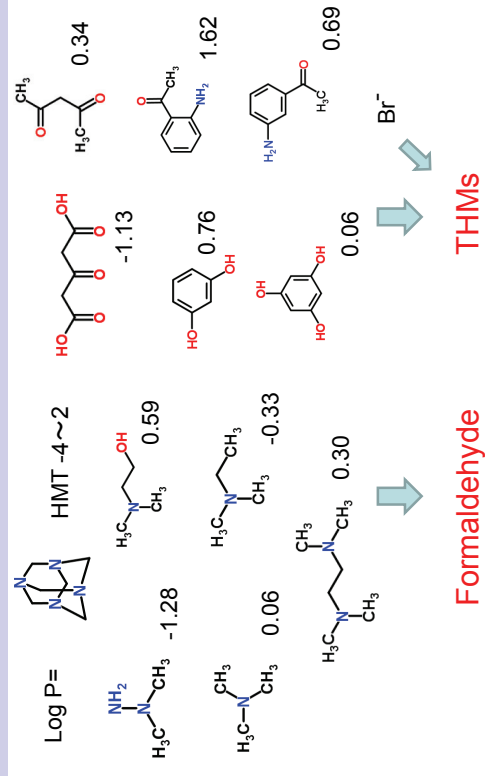
Intake from water is important

Accidental discharge of a formaldehyde precursor into the Tone river in 2012

Formaldehyde was detected above the WQ standard, in the upstream of Tone river. The chemical "Hexamine" was discharged into the Tone river, which was just a 60 tons. It caused water suspension for 850 thousand people.



Submitted a list of troubling chemicals forming DBPs in water treatment





Establishing a database

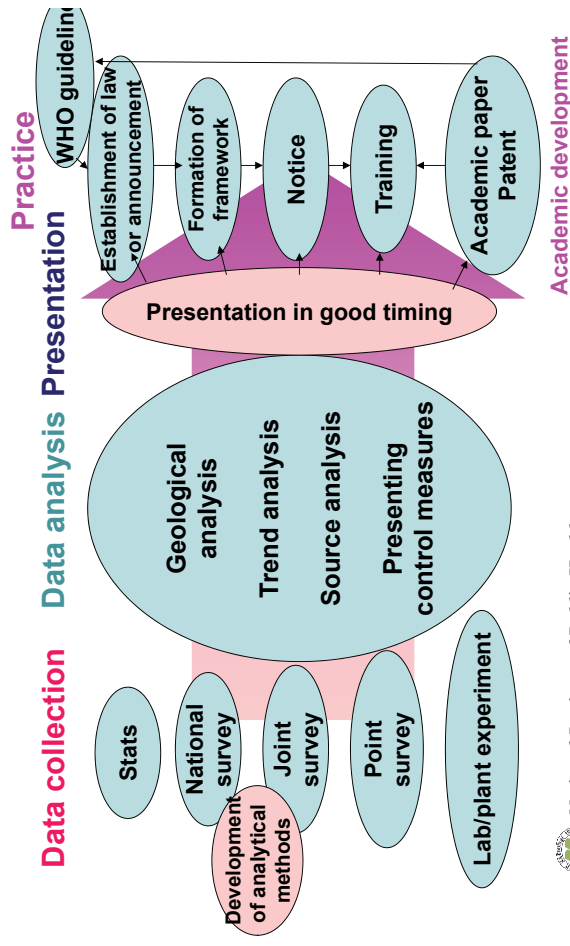
	Log Pow	Log Dow	Henry's law constant
484 108-98-5	パラフィン		
485 7448-01-6	オキサリム酸O-(4-ヒドロキシフェニル)-4-ヒドロキシ- O-(2-ヒドロキシフェニル)-N-ヒドロキシイミド(ヒドロキシ)	<chem>C1=CC=C(C=C1)C(=O)N(C(=O)N2C=CC=C2)C3=CC=C(C=C3)C(=O)N1</chem>	
486 11198-98-5	オキサリム酸O-(4-ヒドロキシフェニル)-4-ヒドロキシ- N-(2-ヒドロキシフェニル)-N-ヒドロキシイミド(ヒドロキシ)	<chem>C1=CC=C(C=C1)C(=O)N(C(=O)N2C=CC=C2)C3=CC=C(C=C3)C(=O)N1</chem>	
487 1183-19-5	パラブロンニゾプロピルニゾプロピル	<chem>CC(C)C1=CC=C(C=C1)C(=O)N(C(=O)N2C=CC=C2)C3=CC=C(C=C3)C(=O)N1</chem>	
488 834-48-5	パラブロン	<chem>CC(C)C1=CC=C(C=C1)C(=O)N(C(=O)N2C=CC=C2)C3=CC=C(C=C3)C(=O)N1</chem>	
489 112-30-1	パラブロンニゾプロピルニゾプロピル	<chem>CC(C)C1=CC=C(C=C1)C(=O)N(C(=O)N2C=CC=C2)C3=CC=C(C=C3)C(=O)N1</chem>	
490 91-11-8	パラブロンニゾプロピルニゾプロピル	<chem>CC(C)C1=CC=C(C=C1)C(=O)N(C(=O)N2C=CC=C2)C3=CC=C(C=C3)C(=O)N1</chem>	

Production, Detection, Toxicity

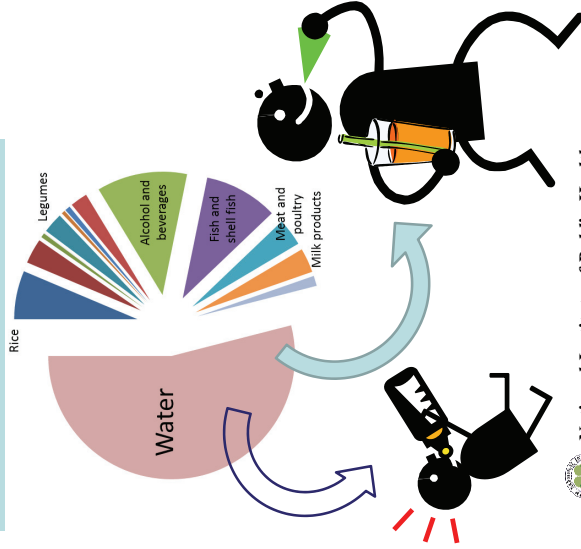
Common data base for chemicals

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Research design for promotion of practice



Total intake of chlorate



Source identification is important!!
 Sometimes water is the main route of exposure of hydrophilic chemicals.

Thank you !

asami@niph.go.jp

For your reference

- Kosaka K, Asami M, Ohkubo K, Iwamoto T, Tanaka Y, Koshino H, Echigo S, Akiba M. Identification of a New N-Nitrosodimethylamine Precursor in Sewage Containing Industrial Effluents. Environ. Sci. Technol. 2014;48 (19):11243-50.
- Asami M, Yoshida N, Kosaka K, Ohno K, Matsui Y. Contribution of tap water to chlorate and perchlorate intake: a market basket study. Science of the Total Environment. 2013;463:199-208.
- Asami M, Kosaka K, and Kunikane S, Bromate, Chlorate, Chlorite and Perchlorate in Sodium Hypochlorite Solution Used in Water Supply, J. Water Supply: Research and Technology AQUA.2009;58(2)107-115.
- Kosaka K, Asami M, Konno Y, Oya M, Kunikane S. Identification of antiyellowing agents as precursors of N-nitrosodimethylamine production on ozonation from sewage treatment plant influent. Environmental Science & Technology 2009; 43(14): 5236-41.